Cleaning Water

5.4

A project of CA NGSS K–8 Early Implementation Initiative.

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Standards
Refer to Appendix 5.4 for NGSS, CCSS ELA and Math, and California ELD standards.

Anchoring Phenomenon
Sewage water is consumed by people, but they do not get sick.

Lesson Concept
Ask questions based on observation, then identify questions that can be investigated to identify patterns in the type and quantity of matter in the town water samples.

Identified Problem
Water collected from neighboring communities may be contaminated. The town officials are requesting help to design a process that will identify the particles in the water and then clean the water.

Standards
Refer to Appendix 5.4 for NGSS, CCSS ELA and Math, and California ELD standards.
Storyline Link

In the last lesson, students learned new properties of matter (magnetism, electrical conductivity, weight, and solubility) and continued to use their knowledge of matter and the various properties of matter to identify substances. These properties can also be useful in solving problems such as separating a mixture.

This lesson centers on students investigating new properties of matter (evaporation and filtration) to gather data and evidence. They use this evidence to plan a design solution for identifying and removing the unwanted materials from the town water samples. Students define criteria and constraints of the water problem as part of the engineering design.

In the next lesson, students will use their possible design solutions to identify and separate matter from the water contained in the town water samples.

Time

3 hours 50 minutes

Part I with optional explore 1 hour 25 minutes + evaporation time

15 minutes Engage
30 minutes Explore 1
30 minutes Optional: Explore 1a
10 minutes Explain 1

Part II 2 hours 25 minutes

60 minutes Explore 2
45 minutes Explain 2
40 minutes Elaborate/Evaluate

Materials

Whole class

❑ Chart paper
❑ Town Water Samples (from Lesson 1: Town Water Samples)
❑ Cups (stored at the end of Lesson 3: Properties of Matter)

Items for making the solution for the filtration system (add other items you might have such as sponges, other fabrics, cotton)
❑ 20 paper bowls
❑ 20 8-oz. clear plastic cups
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- 20 4-in. squares of screens
- 20 4-in. pieces of cheesecloth
- 20 small shallow plastic bowls
- 2 1-quart water containers
- 20 plastic spoons
- 20 1-inch round magnets
- 20 coffee filters
- 1 paper towel roll
- 30 hand lenses
- 12 3-oz. small cups
- 4 small digital kitchen scales
- 15 tablespoons (kitchen measuring type)

Groups (Optional activity in Explore 1) (Groups of 4)

- 4 12-oz. jars
- 2 cups water
- 6 cups granulated sugar
- 4 wooden skewers to hang in jars about an inch from the bottom
- 4 clothespins
- Food coloring

Individual

- Science notebook
- Whiteboard
- Green and yellow sticky notes
- 5.3 H4: Mixing Matter Observations (from Lesson 3: Properties of Matter)
- 5.4.H1: Separating Mixtures
- 5.4.H2: Exit Ticket
- 5.4.H3: Environmental Engineer Design Plan
- 5.4.H4: Sugar Water Assessment
- 5.4.H5: Rubric
Advance Preparation


2. For some specific strategies, explanations, and helpful ideas for doing engineering with fifth graders, review Chapter 11 of the California Science Framework (see references), particularly pages 1478–1484 and 1489–1490. These pages also provide explanations and specifics for the engineering design process.

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Procedure

Part I
Engage (15 minutes)

*Use patterns to ask questions to identify unique properties of matter.*

1. Ask students to discuss with a partner this prompt: “What are some properties of matter that you investigated in our last lesson?” Ask partners to review their data from 5.3.H4: Mixing Matter Observations, paying attention to their observations before and after adding water to each substance. What patterns did they notice? ESRs: some mixed and stayed clear; others did not mix; some looked like they disappeared, but we know that they are still there; the particles are just too small to be seen.

2. Show the cups saved from Lesson 3: Properties of Matter where substances were dissolved in water. Depending on how the solutions were left, there should now be crystals or film residues forming on the craft sticks left in the cups. Have students respond in their science notebook to each of these prompts:
   a. What do you notice about the water in the cups?
   b. How do you think these crystals and residue formed?
   c. What are these crystals and residue?
   d. How can these crystals and residue help us solve the problem of cleaning the contaminated water in the town water samples?
   e. What new questions do you have?

3. After students have sufficient time to think and write in their science notebook, have them share responses with the class. Add new questions to the Design Solutions Question Board.

**TEACHER NOTE**

Focus on the “cleaning the water” section of the Design Solutions Question Board.

Possible Extension: Students may have questions about preventing contamination of water. If so, add that category to the Design Solutions Question Board. If students do not yet have these questions, it is likely that these sorts of questions will emerge soon, at which point the category can be added. Water protection and preservation are not addressed in this learning sequence, but it can be added as an extension or differentiation strategy at the end of the sequence; it addresses 5-ESS3-1.
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Explore 1 (30 minutes)

Obtain information from a text about how the properties of matter are used to identify materials (patterns).

4. Tell students, “Today, we are going to read about some ways to separate different types of matter that are mixed together. This may be helpful in answering some of your questions and provide more ideas for separating the “stuff” in the town water samples and ultimately cleaning the sewage water. As we read, draw a star by examples of mixtures that are separated in the real world. Draw a circle around the names of the processes used to separate mixtures. Underline definitions.”
   a. Hand out 5.4.H1: Separating Mixtures. Read aloud or have pairs do a shared reading.
   b. Discuss the processes of filtration and evaporation in the article and address how the new information could be helpful in developing the plan to clean the water in the jars.

5. Challenge students to identify the difference between the processes of filtration and evaporation by revisiting the definitions in the text. Come to a class consensus on the differences. Encourage students to find evidence in the reading that focuses on the properties of matter that are important in both processes and reference exactly where the information is found in the text. ESRs: Filtration is dependent on particle size, and evaporation is dependent on the types of matter changing phases at different times. Liquid turning into a gas is the phase change important in evaporation (and distillation). Evaporation is dependent on the property of boiling point temperature.

   TEACHER NOTE

   Filtration, distillation, and evaporation are addressed in the reading. At this grade level, distillation and evaporation can be lumped together for the purpose of discussion.

   Students are not expected to know boiling points at this grade level, but they should be able to state that there are properties of water and sugar (or other matter) that result in the materials evaporating at different temperatures.

6. Identify and discuss which process resulted in the formation of the crystals and residue discovered in the cups from Lesson 3: Properties of Matter and examined in Step 2 of this lesson.

   The expected student response is evaporation.
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Explore 1a: Optional (30 minutes, plus overnight for evaporation)

Conduct an experiment of making rock candy to identify the process of evaporation.

TEACHER NOTE

If students have difficulty understanding evaporation, have students create “rock candy” to develop a connection to the process. The directions below will yield 4 rock candy sticks of the same color. You might want to ask what color students want, and then group them by color choice to make their candy. If all of the ingredients are clean, students can eat the product!

Teacher prep: Clean the glass jars thoroughly with hot water; cut wooden skewers so that they hang about 1 inch from the bottom of the jars. Find spring-loaded clothespins that are at least the length of the jar mouth.

1a. Ask each student to wet their wooden skewer with water and roll it in granulated sugar. This base layer gives the sugar crystals something to grab onto when they start forming.

2a. Ask groups of four to create their sugar syrup. Place the water in a medium-sized pan and bring it to a boil. Begin adding the sugar, 1 cup at a time, stirring after each addition. You will notice that it takes longer for the sugar to dissolve after each cup you add. Continue to stir and boil the syrup until all of the sugar has been added, and it is completely dissolved. Remove the pan from the heat.

Note if you are uncomfortable with students doing this, set up a station in the front of the classroom where you can monitor this process.

3a. Have students add 2–3 drops of their favorite food coloring and stir it in to ensure an even, smooth color. Then allow the sugar syrup to cool for 20 to 30 minutes.

4a. Ask students to rinse the prepped jars with hot water, then pour the syrup into them.

5a. Ask each student to use a clothespin as a horizontal holder to lower one sugared skewer into each jar until it hangs about 1-inch from the bottom. It should not touch the bottom of the jar.
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6a. Ask students to carefully place their jar in a cool place, away from harsh lights, where it can sit undisturbed. Cover the top loosely with plastic wrap or a paper towel.

**TEACHER NOTE**
You should start to see sugar crystals forming within 2 to 4 hours. If you see no change after 24 hours, try boiling the sugar syrup again and dissolve another cup of sugar into it. Then pour it back into the jar and insert the skewer again.

7a. Allow the rock candy to grow until it is the size you want. It may take several days. Don’t let it grow too wide, or it might be impossible to remove from the jar.

8a. Note that a top layer of crystal will form. This is okay. Once the candy has reached the desired size, break up that top layer of crystal with a fork before removing the candy.

9a. Transfer the rock candy to an empty jar or glass (keep the clothespins to balance it) and allow it to dry for 1 to 2 hours, then enjoy or wrap in plastic wrap to save for later.

**Explain 1 (10 minutes)**

Communicate information about properties of matter to make predictions based on size.

7. ▶ Students write a response on 5.4.H2: Exit Ticket for each of these questions:
   a. How does understanding properties of matter help us use filtration to identify materials?
   b. How did the crystals and residue form in the cups that were left out from the last lesson?
   c. Predict what would happen if a mixture of sand and water was left to evaporate.
   d. Based on your learning so far, what suggestions would you give city leaders regarding the neighboring town’s drinking water problem?

8. Students discuss their responses and determine if they can add any evidence or new questions to the Class Question Board. Collect the 5.4.H2: Exit Ticket.


**TEACHER NOTE**
Share with students some response groupings such as “I notice that about one-third of the class said _____. And a few said _____.” Or “I noticed that a few people said _____. (some interesting or outlandish response).” Do this for each of the questions. Find responses that stimulate interesting discussions and promote ideas for solutions to determine the materials in the town water samples.
Part II Explore 2 (60 minutes)

Define the problem and design a solution to separate and identify materials based on their properties.

10. Remind the class that they have been hired as environmental engineers to solve a problem. Ask, "What is the problem we are trying to solve? What are some issues to consider when thinking about how to resolve the problem?" 

ESRs: The water is contaminated so we have to think about how to get the contaminants out of the water. We have to make observations and use of the properties of matter to identify and quantify the matter in each water sample, and then a design for a separation process can be made.

11. Distribute 5.4.H3: Environmental Engineer Design Plan and have students record the problem in the box labeled #1. If needed, students can be provided with this sentence stem:

_____ needs a way to _____ so that _____.

ESR: The town needs a way to identify matter in the water so that the contaminants can be separated from the water, making the water safe for drinking.

12. Ask table groups to identify the criteria that must be met in order to successfully resolve the water problem. Ask groups to discuss what successfully solving the problem would look like. Ask groups to share ideas and then as a class, agree to the criteria that will be used to determine a solution to the problem. Ask students to record the agreed-upon criteria in the box labeled #2 on 5.4.H3: Environmental Engineer Design Plan.

• What are some things that are important when we consider drinking water?
• Based on these important things, what would a successful process need to do if the goal was safe drinking water?

ESRs: The visible particles will be removed from the water; no crystals or residue will be left when a small sample of water is evaporated.
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It is helpful to have table groups rank and order the criteria from most important to least important. This ranking can help teams make design decisions later.

**Teacher Note**

Criteria are the specific qualities of a successfully designed solution. For example, strength, durability, reliability, and speed can all be criteria for a design. If students need a reminder, play The Engineering Design Process: A Taco Party video, which compares the engineering design process to hosting a taco party.

13. Ask table groups to determine the limitations that are present in creating a solution to this problem. Display the materials that are available for use. (See materials list.) Provide names for items if students ask but refrain from providing details on ways to use the tools.

14. Ask table groups to share ideas, and then as a whole class agree to the constraints that will be imposed on the design solution. Then have students write the constraints in the box labeled #3 on 5.4.H3: Environmental Engineer Design Plan.

**Teacher Note**

Constraints describe the limitations on a design, such as resources (e.g., time, materials, and funds).

You can set some of the constraints such as Internet use or whether other materials can be used besides the ones you’ve provided. You can impose a maximum funding limit as long as you provide material item costs so that groups are limited to purchasing materials within a fictitious budget.

Ranking the importance of the constraints can also be helpful for group discussions.

15. Students review their exit slip responses and feedback from Step 7 and discuss with their group this prompt: “Thinking about your lesson experiences, what are some ways that your group can solve this problem and use properties of matter to first separate and then identify the matter?” Have them record the scientific information they know in the box labeled #4 on 5.4.H3: Environmental Engineer Design Plan.

16. Ask table groups to design and write a process/plan in the box labeled #5 on 5.4.H3: Environmental Engineer Design Plan. The plan must result in a solution to the #1: problem they identified, which most closely meets #2: the criteria they established within #3: the limitations they identified using the #4: scientific information they learned in the previous lessons. Circulate and encourage groups to address how data will be collected. This data can be qualitative and/or quantitative.
17. After groups have time to brainstorm, have a whole-class discussion of the appropriate types of data to collect. You may want to discuss the benefits of conducting multiple trials in an experiment. Ask, “How can we use math to help describe and measure this scale of filtration?” Facilitate a class discussion on the importance of conducting multiple trials to validate results. ESRs: We can collect data on the amount of water going into the filter and how much is coming out. We can compare the color before and after the filtration process. We can repeat our process multiple times.

18. Ask table groups to make a prediction stating why their plan will work. They should explain why each part of their plan will work to separate substances from the water. Students should justify the order of their steps in terms of why they think it will allow for the separation of each substance. You can provide this sentence frame to guide them:

If I _____, then _____ would happen because _____ has shown to be effective for this phase of the solution to the problem.

**TEACHER NOTE**

Differentiation strategy

If students are familiar with the crosscutting concept Systems and System Models, then a sketch of their process can be drawn as a system, the components labeled, and the interactions described instead of writing out the specifics for each part of their system.

**TEACHER NOTE**

By the end of grade 5, students are expected to have the skill to “respectfully provide and receive critiques from peers about a proposed procedure, explanation, or model by citing relevant evidence and posing specific questions.”

For more information on this SEP, read chapters 11 and 13 of *Helping Students Make Sense of the World Using Next Generation Science and Engineering Practices* by Christina Schwarz, Cynthia Passmore, and Brian J. Reiser.
19. Conduct a table group feedback review of the plans (similar to a gallery walk). One group member remains at the table to present the plan to visiting classmates and receive feedback/questions while other group members visit different groups to provide feedback on their proposed procedures.

20. Direct students to use green sticky notes to identify where the plans are valid and to use yellow sticky notes to write probing questions where the plans need more thought. Circulate, providing feedback and asking probing questions as needed.

21. Hand out 5.4.H5: Rubric and tell students to consider only the first two rows at this time (defining problems and developing and using models). Ask table groups to reconvene, consider their peer feedback, and use the descriptions on the rubric to modify their plan.

22. Tell table groups that they will address the rest of the rubric and complete the remaining boxes on 5.4.H3: Environmental Engineer Design Plan in the next lesson.

Elaborate/Evaluate (40 minutes)

Communicate information including a model for a design solution to identify materials in a sugar water solution.

23. Ask students to write in their science notebook a reflection on what they have learned about the engineering design process. If they have any new questions, ask them to write them on a sticky note and place them on a parking lot poster.

24. Add the agreed-upon problem, criteria, and constraints to the third section of the Design Solutions Question Board (cleaning the water). In Lesson 5: Separating Mixtures, the student-designed plans will be tested and results will be added as final pieces of evidence.

25. Have students individually complete 5.4.H4: Sugar Water to demonstrate their understanding of:
   a. a small quantity of matter (sugar) existing in a larger quantity of other matter (water)
   b. how to draw a model demonstrating their understanding
   c. designing a process to provide evidence

26. Write the letters A-F on the board, leaving space to record the numbers of students who chose each. Then ask those who chose A to stand and choose one or two to explain their reasoning. Repeat for all answers thru F. (The correct answers are C, D, & F). Explain to students that in grades 6–8 they will continue to build their understanding about what happens to substances when they are dissolved in a mixture.
27. Collect 5.4.H4: Sugar Water and review the student responses. The responses will provide you with information as to what the students still do not understand about mixtures.

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Separating Mixtures

A mixture is a substance made by combining two or more different materials. The materials maintain their original properties when mixed in such a way that no chemical reaction occurs. A mixture can usually be separated back into its original components. Some examples of mixtures are a tossed salad, a basket of different fruits and vegetables, a toy box filled with toys, and dust in the air.

In the gold rush days in California, miners separated gold from the mixture of gold, dirt, water, and sand in streams and rivers. The gold was heavier than the sand and the dirt, so the miners developed different ways of getting the gold to settle to the bottom of their device while washing away the dirt and sand.

There are many different ways that mixtures can be separated. Three of the separation processes are filtration, distillation, and evaporation.

One method for separation is filtration. It uses a filter such as paper or sand to allow a liquid or gas to pass through. This traps the solids or impurities on the filter. When you make coffee in the morning, the grinds (solid) are trapped by the filter, as the water (liquid) passes through the beans to make the coffee. There are many types of filters. Some are also used to trap dust in our homes and businesses. Others remove impurities from our drinking water. Face masks filter bacteria and germs so we don’t breathe them in. Even our bodies have filters—for example, the hairs in our nasal passages prevent impurities from reaching our lungs. Our kidneys and liver filter the impurities in our blood.

The filtration process is the process by which a mixture (small solid particles in water or air) are separated. When filtering water, the water is forced through paper or another material that is made up of a tiny fine mesh of fibers. This lets the water can pass through while trapping the small particles. The particles removed from the water by the filter are called the residue.

Distillation, another method for the separation process, uses boiling to separate liquid mixtures. Different substances often have different boiling points, which means they boil at different temperatures. For example, water boils at 100°C, and linseed oil boils at 287°C. If you heat saltwater to boiling, the water will turn into steam and leave the salt crystals behind. If the steam from the boiling salt water is allowed to cool, it will form droplets of fresh water, without salt. One example of distillation is converting saltwater into drinking water. Another example of distillation is heating crude oil to separate it into its many components such as jet fuel, engine oil, and gasoline.

Evaporation is the process by which a liquid changes from a liquid to a gas (vapor). Some examples of evaporation are wet clothes drying in the sun on a clothesline, drying of wet hair, and hot tea getting cold. Our bodies use the evaporation of sweat to cool us. Removing salt from ocean water can be achieved through distillation AND through the evaporation process.
Exit Ticket

1. How does understanding properties of matter help us to use filtration to identify materials?

2. How did the crystals and residue form in the cups that were left out from the last lesson?

3. Predict what would happen if a mixture of sand and water was left to evaporate.

4. Based on your learning so far, what suggestions would you give city leaders regarding the town’s drinking water problem?
Environmental Engineer Design Plan

(for cleaning water samples)

1. Identify the problem.

The problem is

2. Define what you are asked to do (the criteria).

Our environmental engineer team has been asked to

3. Define the limitations or requirements (the constraints) necessary to solve the problem.

4. Apply scientific information have you learned from previous lessons and the reading to help you solve the problem.
Environmental Engineer Design Plan (continued)

5. Create a plan. Write detailed steps.

6. Revise your plan.
Environmental Engineer Design Plan (continued)

7. Carry out your plan and record your observations and results. You will use these results to make a claim.
   Observations and results:

8. Compare your observations and results with those of another group. Which plan or parts of your plan worked more efficiently? Which parts did not work well?
   Our plan met the criteria or was efficient at ___________________based on these observations:

   Our plan did not meet the criteria or was less efficient at ___________________ based on these observations:
Sugar Water

Deanna stirred a teaspoon of sugar (5 grams) into a glass of warm water (200 grams). The sugar completely dissolved in the water. Put an "X" next to the statements that are true about the dissolved sugar.

- A The sugar melts.
- B The sugar turns into water particles.
- C The sugar forms a mixture with the water.
- D The sugar can be separated from the water.
- E The sugar disappears and no longer exists.
- F The sugar particles are spread among the water particles.

1. Draw a model to explain what happens when sugar dissolves in water. Include both the visible and invisible parts, labels, and amount.
2. Based on the investigations you have completed, briefly describe a plan to provide evidence that sugar is dissolved in the water.

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Adapted from Page Keeley and Joyce Tugel, Uncovering Student Ideas in Science Vol. 4.
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<td><strong>Defining problems</strong></td>
<td>We understood the design problem.</td>
<td>We understood the design problem.</td>
<td>We had a good understanding of the design problem.</td>
<td>We had an exceptional understanding of the design problem and could clearly articulate it. We paid attention to multiple criteria and constraints and understood the relationships between them. We used science knowledge and could clearly explain the science behind our design.</td>
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<tr>
<td>Define a design problem that can be solved through the development of an object, tool, process or system and includes multiple criteria and constraints.</td>
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<tr>
<td><strong>Developing and using models</strong></td>
<td>We created a model of what we thought was in the jars. For example, we saw the jars, and we drew a picture of the jars with some labels.</td>
<td>We created a model of the jars and revised it to show what was really in the jars. We added labels.</td>
<td>We created and revised a model of our jars. We understood and could name all the substances. Our model accurately represents the relationships among the parts of the proposed solution.</td>
<td>We created and revised a model of our jars. We could identify all the substances and their relationships within the solution. Our model explained the relationship between each of the parts and addressed the accuracy and limitations of the model.</td>
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<tr>
<td>Develop or modify a model—based on evidence—to explain the phenomena</td>
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<tr>
<td><strong>Planning and carrying out investigations</strong></td>
<td>We collected qualitative performance data of the tools used to filter the solutions for a single trial.</td>
<td>We collected qualitative performance data of the tools used to filter the solutions for a single trial. We evaluated the accuracy of some of the tools used to filter the solutions.</td>
<td>We collected quantitative and qualitative performance data of the tools used to filter the solutions using multiple trials. We evaluated the accuracy of the tools used to filter the solutions.</td>
<td>We collected quantitative and qualitative performance data of the tools used to filter the solution using multiple trials and models of the tools. We evaluated the accuracy of different models of the same tool used to filter the solutions.</td>
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<tr>
<td>Collect data about the performance of a proposed object. Evaluate the accuracy of various methods for collecting data.</td>
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<tr>
<td><strong>Analyzing and interpreting data</strong></td>
<td>We analyzed the data of the single trial.</td>
<td>We analyzed the performance data of the single trial.</td>
<td>We analyzed the performance data from multiple trials.</td>
<td>We analyzed the performance data from multiple trials to refine the design of the tools used to filter the solutions.</td>
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<tr>
<td>Analyze data to define an optimal process or system that best meets criteria for success. Consider limitations of data analysis (e.g., measurement error), and/or seek to improve precision and accuracy of data and methods (e.g., multiple trials).</td>
<td></td>
<td>We considered the limitations of our data collection.</td>
<td>We considered the limitations of our data collection and analysis.</td>
<td>We created and/or used different tools to improve the precision and accuracy of the data.</td>
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<tr>
<td><strong>Using mathematics and computational thinking</strong></td>
<td>We found a good solution to the design problem using trial and error.</td>
<td>We used some digital tools and/or mathematical concepts to qualitatively evaluate a proposed design solution.</td>
<td>We used appropriate quantitative mathematical tools and/or concepts to design and evaluate the proposed solution.</td>
<td>We used appropriate mathematical concepts to design and quantitatively evaluate different tools used to solve the same problem.</td>
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### Rubric (continued)

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<td>We proposed a design solution that meets some criteria and constraints.</td>
<td>We proposed a design solution that meets the criteria and constraints and is Based on scientific ideas.</td>
<td>We proposed a design solution that meets the criteria and constraints, and using data, clearly supports the scientific basis for the design using data.</td>
<td>We proposed a design solution that meets the criteria and constraints and compares supporting data from more than one proposed design solution.</td>
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Appendix 5.4

Cleaning Water

Next Generation Science Standards (NGSS)

This lesson is building toward:

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<tr>
<td>Make observations and measurements to identify materials based on their properties. [Clarification Statement: Examples of materials to be identified could include baking soda and other powders, metals, minerals, and liquids. Examples of properties could include color, hardness, reflectivity, electrical conductivity, thermal conductivity, response to magnetic forces, and solubility; density is not intended as an identifiable property.] [Assessment Boundary: Assessment does not include density or distinguishing mass and weight.]</td>
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<tr>
<td><strong>3-5 ETS1-1</strong></td>
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<td>Define a simple design problem reflecting a need or a want that includes specified criteria for success and constraints on materials, time, or cost.</td>
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<td><strong>Asking Questions and Defining Problems</strong></td>
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<td>• Ask questions that can be investigated and predict reasonable outcomes based on patterns such as cause and effect relationships.</td>
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<td>• Use prior knowledge to describe problems that can be solved.</td>
</tr>
<tr>
<td>• Define a simple design problem that can be solved through the development of an object, tool, process, or system and includes several criteria for success and constraints on material, time, or cost.</td>
</tr>
<tr>
<td><strong>Planning and Carrying Out Investigations</strong></td>
</tr>
<tr>
<td>• Make observations and /or measurements to produce data to serve as the basis for evidence for an explanation of a phenomenon or to test a design solution.</td>
</tr>
<tr>
<td><strong>Analyzing and Interpreting Data</strong></td>
</tr>
<tr>
<td>• Analyze and interpret data to make sense of phenomena, using logical reasoning, mathematics, and /or computation.</td>
</tr>
<tr>
<td>• Compare and contrast data collected by different groups in order to discuss similarities and differences in their findings.</td>
</tr>
<tr>
<td><strong>Constructing Explanations and Designing Solutions</strong></td>
</tr>
<tr>
<td>• Apply scientific ideas to solve problems.</td>
</tr>
<tr>
<td>• Use evidence (e.g., measurements, observations, patterns) to construct of support an explanation or design a solution to a problem.</td>
</tr>
<tr>
<td><strong>Obtaining, Evaluating, and Communicating Information</strong></td>
</tr>
<tr>
<td>• Read and comprehend grade-appropriate complex texts and/or other reliable media to summarize and obtain scientific and technical ideas and describe how they are supported by evidence.</td>
</tr>
<tr>
<td>• Communicate scientific and/or technical information orally and/or in written formats, including various forms of media, as well as table, diagrams and charts.</td>
</tr>
</tbody>
</table>
Appendix 5.4

DISCIPLINARY CORE IDEAS (DCI)


• Measurements of a variety of properties can be used to identify materials.
• Matter of any type can be subdivided into particles that are too small to see, but even then, the matter still exists and can be detected by other means.

ETS1.A: Defining and Delimiting Engineering Problems

• Possible solutions to a problem are limited by available materials and resources (constraints). The success of a designed solution is determined by considering the desired features of a solution (criteria). Different proposals for solutions can be compared on the basis of how well each one meets the specific criteria for success or how well each takes the constraints into account.

ETS1.B: Developing Possible Solutions

• At whatever stage, communicating with peers about proposed solutions is an important part of the design process, and shared ideas can lead to improved designs.

CROSSCUTTING CONCEPTS (CCC)

Patterns

• Identify similarities and differences in order to sort and classify natural objects and designed products.
• Identify patterns related to time, including simple rates of change and cycles and use these patterns to make predictions.

Scale, Proportion, and Quantity

• Natural objects and/or observable phenomena exist from the very small to the immensely large or from very short to very long time periods.

Influence of Engineering, Technology, and Science on Society and the Natural World

• People’s needs and wants change over time, as do their demands for new and improved technologies.

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Common Core State Standards (CCSS)

CCSS ELA WRITING

CCSS.ELA-LITERACY.W.5.8
Recall relevant information from experiences or gather relevant information from print and digital sources; summarize or paraphrase information in notes and finished work, and provide a list of sources.

CCSS.ELA-LITERACY.W.5.9
Draw evidence from literary or informational texts to support analysis, reflection, and research.
## Appendix 5.4

### MATHEMATICS PRACTICES

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<tr>
<td>MP.2</td>
<td>Reason abstractly and quantitatively.</td>
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<tr>
<td>MP.5</td>
<td>Use appropriate tools strategically.</td>
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### California English Language Development (ELD) Standards

**CA ELD**

**Part 1.5.3** Offering opinions and evaluating others’ opinions in speaking and writing.

<table>
<thead>
<tr>
<th>EMBEDDING</th>
<th>EXPANDING</th>
<th>BRIDGING</th>
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<tbody>
<tr>
<td>Negotiate with or persuade others in conversations using basic learned phrases (e.g., <em>I think . . .</em>), as well as open responses, in order to gain and/or hold the floor.</td>
<td>Negotiate with or persuade others in conversations using an expanded set of learned phrases (e.g., <em>I agree with X, but . . .</em>), as well as open responses, in order to gain and/or hold the floor, provide counter-arguments, etc.</td>
<td>Negotiate with or persuade others in conversations using a variety of learned phrases (e.g., <em>That's an interesting idea. However . . .</em>), as well as open responses, in order to gain and/or hold the floor, provide counter-arguments, elaborate on an idea, etc.</td>
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In addition to the standard above, you may find that you touch on the following standard as well:

**P1.5.1** Exchanging information and ideas with others through oral collaborative discussions on a range of social and academic topics.

**P1.5.2** Reading closely informational texts and viewing multimedia to determine how meaning is conveyed explicitly and implicitly through language.

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